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Molar gas volume = 24.0 dm³ mol⁻¹ at room temperature and pressure, RTP

Avogadro constant, N_A = 6.02 × 10²³ mol⁻¹

Specific heat capacity of water, c = 4.18 J g⁻¹ K⁻¹

Ionic product of water, K_w = 1.00 × 10⁻¹⁴ mol² dm⁻⁶ at 298 K

1 tonne = 10⁶ g

Arrhenius equation: k = Ae^{-E_a/RT} or ln k = -E_a/RT + ln A ...

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